

# Prediction of mixed hardwood lignin and carbohydrate content using ATR-FTIR and FT-NIR



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## ABSTRACT

This study used Attenuated total reflectance Fourier transform infrared (ATR-FTIR) spectroscopy and Fourier transform near-infrared (FT-NIR) spectroscopy with principal component regression (PCR) and partial least squares regression (PLS) to build hardwood prediction models. Wet chemistry analysis coupled with high performance liquid chromatography (HPLC) was employed to obtain the chemical composition of these samples. Spectra loadings were studied to identify key wavenumber in the prediction of chemical composition. NIR-PLS and FTIR-PLS performed the best for extractives, lignin and xylose, whose residual predictive deviation (RPD) values were all over 3 and indicates the potential for either instrument to provide superior prediction models with NIR performing slightly better. During testing, it was found that more accurate determination of holocellulose content was possible when HPLC was used. Independent chemometric models, for FT-NIR and ATR-FTIR, identified similar functional groups responsible for the prediction of chemical composition and suggested that coupling the two techniques could strengthen interpretation and prediction.

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## 1. Introduction

As the most abundant fibrous material, wood is utilized in numerous areas including textiles (Mak, Yuen, Ku, & Kan, 2006), paper making (Weinstock et al., 1997), building construction (Ellingwood, Rosowsky, Li, & Kim, 2004), composites (Ruffing, Shi, Brown, & Smith, 2011) and bioenergy (Celikbag, Via, Adhikari, & Wu, 2014). The key bio-based polymers for wood are cellulose, hemicellulose and lignin. The chemical composition of wood plays an important role when one evaluates the utility of a feedstock for various product streams. Rapid assessment of biomass would allow for better categorization of raw materials for end use, allow for better process control, and assist in genetic selection of superior trees. Wet chemistry methods are currently still preferred to yield accurate and precise results. But wet methods are complex

and consume many chemicals over long periods of time. Moreover, chemical modification of the fiber can destroy the wood matrix which is unacceptable for some studies. The slower time to testing also limits the sample size that one can measure in a short period of time (Ferraz, Baeza, & Duránt, 1991; Jiang, Han, Zhang, & Wang, 2010; Kaar & Brink, 1991).

Fourier transform near-infrared (FT-NIR) spectroscopy possesses unique advantages, such as being fast, nondestructive, simple and low in cost (Roggo et al., 2007; Tsuchikawa, 2007). More recently, FT-NIR models were found to be suitable for the analysis of the chemical composition of wood (Chen et al., 2010; Raymond & Schimleck, 2002; Via, Shupe, Groom, Stine, & So, 2003), corn stover (Philip Ye et al., 2008), rice straw (Jin & Chen, 2007) and tobacco (Zhang, Cong, Xie, & Zhao, 2008). These studies utilized both principal component regression (PCR) and partial least squares regression (PLS) to calibrate to the NIR spectra. It was found that PLS performed better than PCR when dealing with low detection-limit samples (Lipp, 1996), while PCR could yield simple and better models for cases where interpretation of the model was important (Jouan-Rimbaud, Walczak, Massart, Last, & Prebble, 1995; Via, Zhou, Acquah, Jiang, & Eckhardt, 2014).

Like FT-NIR, FTIR has the potential to predict with high accuracy and precision but often comparisons are limited (Chen et al., 2010; Schultz & Burns, 1990). For pharmaceutical applications, FTIR

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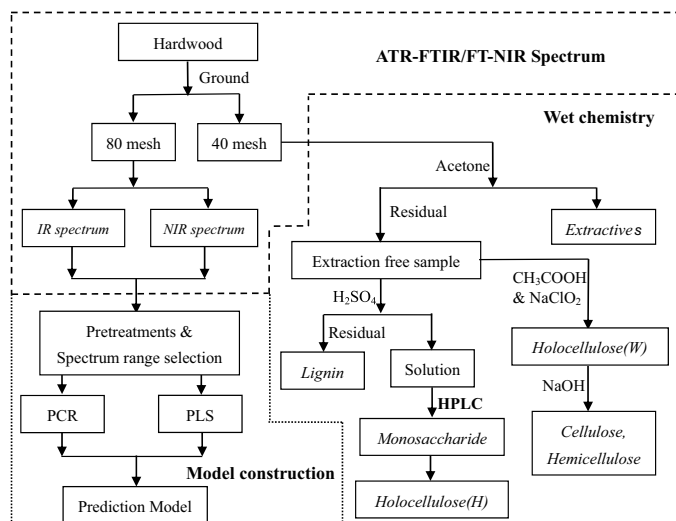


Fig. 1. Schematic diagram for wet chemistry and ATR-FTIR/FT-NIR model construction.

was found to perform better than FT-NIR for the classification of five different cellulosic materials (Langkilde & Svantesson, 1995). It was also demonstrated that FT-IR performed better for the determination of soil properties (Rossel, Walvoort, McBratney, Janik, & Skjemstad, 2006). To date, the comparison of techniques for wood based materials is limited to studies that utilize less robust multivariate techniques (Schultz & Burns, 1990).

This research aimed to evaluate the ability to build calibration equations for ATR-FTIR and FT-NIR using different regression techniques for the hardwood chemistry of several commercially important genus/species. Interpretation of PLS and PCR loadings was also performed to assess model quality.

## 2. Experimental

### 2.1. Sample preparation

All samples were collected from newly harvested hardwood trees. There were 4 different genus/species to generate 37 samples in which 4 Eucalyptus, 9 cotton wood, 12 aspen and 12 poplar were used for testing. First, the solid wood samples were planed down to 3 mm thick wood chips (about 100 g), and then placed in ambient conditions (23–26 °C and less than 50% humidity) for 2 weeks to evaporate free water within the cell lumen. Then 50 g of the air dried samples were ground to 40 mesh using a Willey mill and then 20 g of 40 mesh samples were further ground to 80 mesh. The 40 mesh samples were used for wet chemistry analysis and 80 mesh samples were used for ATR-FTIR and FT-NIR spectra collection. Grinding these samples to a fine mesh was considered important since the porosity of solid materials can influence the reflectance of light (Dutta, Raghavan, & Ngadi, 2012) resulting in higher errors during prediction (Jiang et al., 2014a; Jiang, Via, Han, Wang, & Liu, 2014b).

### 2.2. Wet chemistry analysis

The extractives, lignin, and monosaccharide content of 37 samples were measured on the basis of standard procedures developed by the National Renewable Energy Laboratory (NREL) (Sluiter et al., 2008). The cellulose, hemicellulose, and holocellulose contents were also measured by traditional wet chemistry analysis. As shown in Fig. 1, 150 mL acetone was used to extract 5 g of sample for 6 h to get acetone extractives. After extractive removal, the extractive free sample was separated into 2 batches.

Batch 1: A 72% (w/w) sulfuric acid treatment at 30 °C for 2 h was used to prehydrolyze the extractive free sample. The solution was then diluted to 4% sulfuric acid with distilled water, sealed in a bottle, and placed in an autoclave for 1 h at 121 °C, and then the residual from the bottle was filtered and oven dried to measure the lignin content. The extractives and lignin contents were measured by gravimetric analysis. To determine the monosaccharide composition (glucose, xylose, etc.), an HPLC (Shimadzu LC-20A), equipped with an Aminex 87 P column and differential refractive index detector, was employed to analyze the sugar solution. Holocellulose content (Holo-HPLC) was calculated as the sum of all the monosaccharides contents.

Batch 2: Delignification treatments were conducted to determine the holocellulose content. The delignification procedure was as follows. First, 2 g was weighed separately and placed into conical flasks (500 mL) with 320 mL distilled water in each flask. Second, the sample was placed in flasks and the flasks were then placed into a water bath (75 °C), and then 1 mL of acetic acid (ACS grade, 100%) and 20 mL 15% (w/w) sodium chlorite were added into each flask every 1 h. After 4 h water bath exposure, the residues were filtered with filter paper and then oven dried for 3 h to test the holocellulose content. After the holocellulose content was determined, 1.5 g of oven dried holocellulose was placed into a 250 mL conical flask. One-hundred milliliter of 17.5% sodium hydroxide was then added and the air in the flask was replaced with nitrogen and the flask was sealed with aluminum foil. Following the flask was placed in a water bath at 20 °C and stirred occasionally until the reaction was complete. Then the solution was filtered through a pre-weighed filter paper and washed with distilled water (at least 500 mL). The sample was then oven dried at 105 °C overnight and weighed. The residue was determined as cellulose and the hemicellulose content was considered to be the difference in holocellulose and cellulose.

When conducting wet chemistry, all samples were air dried and tested for moisture content to calculate the dry weight of the original samples. All experiments were performed in duplicate. All chemicals were purchased from the VWR Company and they were all analytically pure or even chromatographically pure.

### 2.3. ATR-FTIR and FT-NIR acquisition

Spectra from both instruments were acquired from 80 mesh wood samples and all samples were oven dried overnight prior to analysis. For each sample, the wood powder was placed on the diamond (ATR-FTIR) or light path (FT-NIR) and the reflectance spectra were collected. The PerkinElmer spectrum 400 FT-IR/FT-NIR spectrometer was utilized to collect the mid and near infrared spectra. For FT-NIR, the spectrum covered a range of 10,000–4000  $\text{cm}^{-1}$  with a spectral resolution of 4  $\text{cm}^{-1}$ . Each spectrum was collected from an average of 32 scans. For the ATR-FTIR, the spectrum covered a range of 4000–650  $\text{cm}^{-1}$  with a spectral resolution of 4  $\text{cm}^{-1}$ . The pressure of the ATR-FTIR acquisition was  $70 \pm 2$  psi. Each spectrum was collected from an average of 4 scans.

### 2.4. Chemometric analysis

Spectrum Quant+ software was used for model construction. PCR and PLS were the two statistical methods used to build ATR-FTIR and FT-NIR models. These methods allowed for the optimal estimation of the y value from the spectral data matrix. These two modeling methods were further combined with different pre-treatments such as first derivative (FD), offset, flat, multiplication scattering correction (MSC) and standard normal variate (SNV). Thirty-two samples were used to construct models and 5 samples were used for validation. Leave one out cross validation was also ran to ensure similar results to the validation data set. The predictive performance of the models in this paper was evaluated by several

**Table 1**  
Statistical summary of components in wood analyzed with wet chemistry and HPLC methods.

| Chemical composition   | Mean value of content (%) | Range of content (%) | SD value |
|------------------------|---------------------------|----------------------|----------|
| Extractives            | 2.11                      | 0.66–5.93            | 1.42     |
| Lignin                 | 26.24                     | 20.85–31.63          | 2.41     |
| Cellulose              | 43.71                     | 39.84–49.05          | 2.10     |
| Hemicellulose          | 35.97                     | 28.35–43.12          | 3.88     |
| Holocellulose          | 79.68                     | 71.52–84.66          | 3.49     |
| Holo-HPLC <sup>a</sup> | 64.52                     | 57.85–68.64          | 2.37     |
| Glucose                | 44.34                     | 40.90–47.06          | 1.31     |
| Xylose                 | 16.69                     | 13.76–20.31          | 1.56     |
| Total (W) <sup>b</sup> | 108.03                    | 101.49–113.07        | N/A      |
| Total (H) <sup>c</sup> | 92.88                     | 87.36–98.38          | N/A      |

<sup>a</sup> Holo-HPLC represents the holocellulose content calculated from HPLC results.

<sup>b</sup> Total (W) represents the total content that was calculated from wet chemistry. Total (W) = extractives + lignin + holocellulose.

<sup>c</sup> Total (H) represents the total content that was calculated from HPLC. Total (H) = extractives + lignin + holo-HPLC.

**Table 2**  
RMSEP value of components in wood analyzed with different methods.

| Chemical composition | FT-NIR |      | ATR-FTIR |      |
|----------------------|--------|------|----------|------|
|                      | PCR    | PLS  | PCR      | PLS  |
| Extractives          | 0.80   | 0.37 | 0.69     | 0.49 |
| Lignin               | 0.87   | 0.46 | 0.83     | 0.84 |
| Cellulose            | 2.00   | 1.79 | 4.03     | 3.20 |
| Hemicellulose        | 1.53   | 0.97 | 1.61     | 2.00 |
| Holocellulose        | 1.94   | 1.69 | 2.43     | 2.60 |
| Holo-HPLC            | 1.59   | 1.29 | 1.76     | 1.61 |
| Glucose              | 0.85   | 0.63 | 0.69     | 0.90 |
| Xylose               | 0.72   | 0.46 | 0.68     | 0.62 |

standards, including correlation coefficient ( $r$ ), root mean square error of calibration (RMSEC), root mean square error of prediction (RMSEP) and residual predictive deviation (RPD).

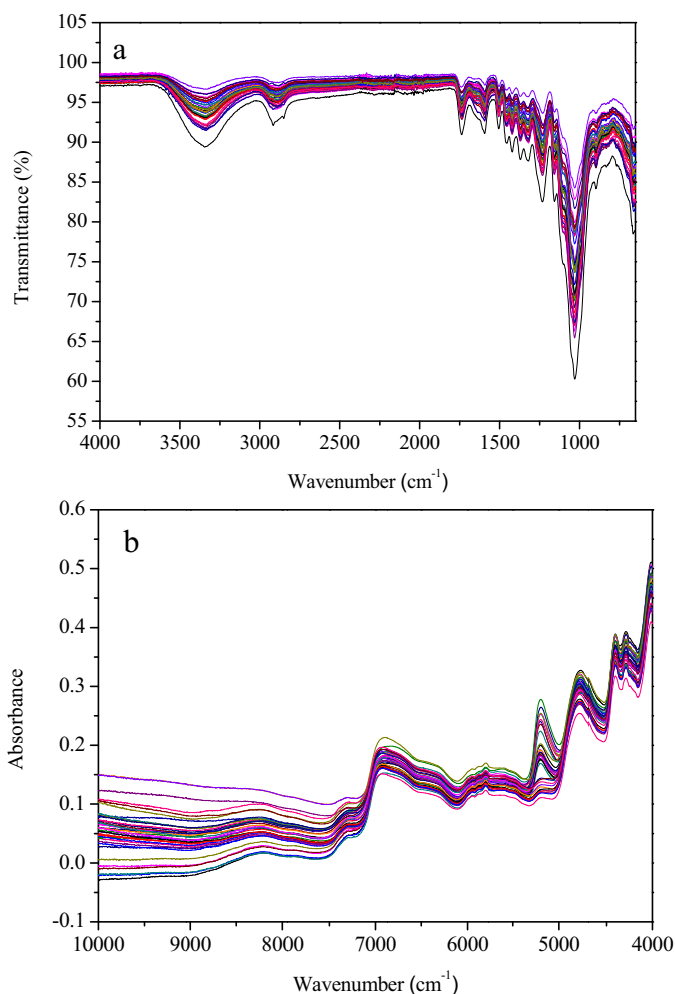
$$RPD = \sqrt{\frac{\sum_{i=1}^n (C_i - \bar{C}_i)^2}{n-1}} / RMSEP. \quad (1)$$

where  $C_i$  is the wet chemistry value in the validation;  $\bar{C}_i$  is the mean of the wet chemistry value in the validation;  $n$  is the number of samples in the validation. In this research, the RPD value was calculated from Tables 1 and 2. A good model should have a high  $r$ , a high RPD, and a low RMSEC, RMSEP.

### 3. Results and discussion

#### 3.1. Wet chemistry and HPLC analysis results

Two methods (Batch 1: HPLC and Batch 2: wet chemistry) were used to determine the carbohydrates content. The results for both methods were found to be systematically different during the measurement of holocellulose. As shown in Table 1, after using wet chemistry analysis, the range in lignin content was 20.85–31.63%, cellulose content (39.84–49.05%), hemicellulose content (28.35–43.12%) and holocellulose content (71.52–84.66%). These findings were statistically similar to other studies demonstrating a representative sample (Pettersen, 1984). However, the range of the total (W) content (101.49–113.07%) was higher than 100%, which implied that some experimental error existed. The error was considered to be some portion of lignin which was not completely removed in the delignification treatment (Kačák, Ďurkovič, & Kačiková, 2012). The lignin was counted as holocellulose, which resulted in the inaccuracy of holocellulose content. More recently, HPLC was widely used to analyze carbohydrates,



**Fig. 2.** ATR-FTIR (a) and FT-NIR (b) spectra used for the predictions.

but minute degradation of some monosaccharides during HPLC processing can occur resulting in a mass balance below 100% (Li & Xu, 2013; Oliver, Gaborieau, Hilder, & Castignolles, 2013). Moreover, there are a few proteins and uronic acids that commonly exist (Sluiter et al., 2008), which were unidentifiable with our HPLC equipment and resulted in Total (H) (87.36–98.38%) values less than 100% (Table 1). Nevertheless, the holocellulose that was determined from HPLC was hypothesized to be closer to the real value than that from wet chemistry. In this research, both holocellulose content (holocellulose/Holo-HPLC) and other carbohydrate contents (hemicellulose, cellulose, glucose and xylose) were used for NIR model training.

#### 3.2. ATR-FTIR and FT-NIR multivariate models

As described above, multivariate models for wood chemical composition were established from ATR-FTIR and FT-NIR spectra (Fig. 2a and b) coupled with PLS and PCR algorithms. This resulted in 4 different prediction models for each chemical component (Figs. 3 and 4). As shown in Figs. 3 and 4, the PLS model coupled with FT-NIR spectroscopy almost always showed the best prediction accuracy, except for xylose in which ATR-FTIR-PLS was statistically similar in  $r$  value. Schultz and Burns (1990) also found NIR based models to provide better prediction errors than FTIR when KBr pellets were used. Similar results were also found when predicting feedstock strand density in which FT-NIR performed better than ATR-FTIR (Via, 2010). The  $r$  values for all components of the

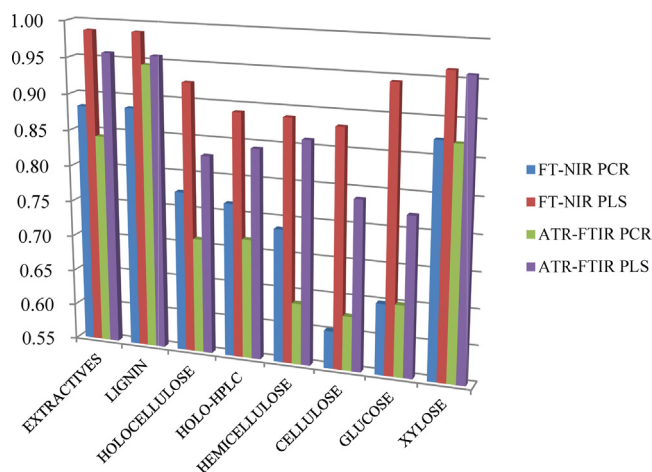


Fig. 3.  $r$  Values for each chemistry component in wood that was analyzed using FT-NIR and ATR-FTIR and different modeling methods.

FT-NIR-PLS model were over 0.85; and the RPD values were all greater than 1.5. RPD values in this range suggest that FT-NIR-PLS can be used for screening purposes during quality control of individual samples (Yeh et al., 2005). However, individual samples are generally not the primary metric of interest during production, but instead population parameters such as production mean and variance are more important indicators of process capability (Via, Adhikari, & Taylor, 2013). These parameters can be estimated with accuracy despite only modest RPD values (Sjoblom, Johnsson, & Sundstrom, 2004).

The best predictive results for the FT-NIR-PLS model occurred for extractives, lignin and xylose in which the  $r$  values for these three models were greater than 0.95, and the RPD values were higher than 3. The greater RPD values for these prediction models suggest that FT-NIR may be more useful than just screening. Other researchers support that models were considered good enough to monitor the quality of individual samples when the  $r$  was  $>0.90$  and the RPD was  $>3$  (Fagan, Everard, & McDonnell, 2011; Williams & Sobering, 1993). Of these two, lignin based models exhibited the best performance in which 96% of the overall variation was accounted for during calibration and this predictive capacity held up during validation with minimum degradation in RMSEP.

The higher RPD values suggest that FT-NIR-PLS prediction models for extractives, lignin and xylose can be used for quantitative analysis of specific samples in a laboratory or manufacturing

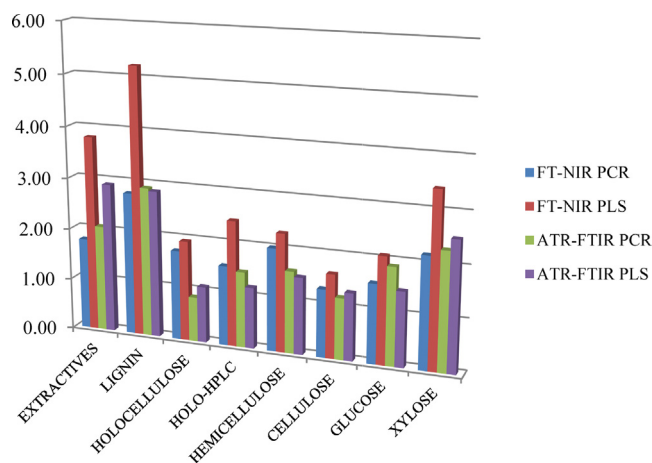


Fig. 4. RPD values for each chemical component in wood that was analyzed using FT-NIR and ATR-FTIR and different modeling methods.

setting. The good prediction diagnostics for extractives and lignin content could be partially explained by their considerable difference from carbohydrates. Carbohydrates, including cellulose, hemicellulose and holocellulose form polysaccharides which contain monosaccharides such as glucose, xylose, and other sugars. These sugars are more similar than they are different with only slight variations in the location of H and OH groups with respect to the cyclical 6 carbon hexagonal monomer structure. And this may result in error when trying to differentiate between different sugar types using chemometric models on ATR-FTIR or FT-NIR spectra.

This study found that PCR performed similarly to their study in which they used MLR models to account for 73% of the variation in hemicellulose concentration. But when PLS was used in this study, the models accounted for 88% of the variation during calibration. Likewise, when the remaining models were evaluated, PLS consistently accounted for more variation in wood chemistry than in PCR. The RPD values were also consistently the highest for PLS for both ATR-FTIR and FT-NIR systems (Fig. 3). However, only the FT-NIR models coupled with PLS achieved RPD values greater than 3, while the remaining models and respective chemical constituents were below 3 and could thus only be useful for screening.

Most models built for ATR-FTIR were below 2 and some even less than 1.5. These findings support that ATR-FTIR based models generally fall within the screening capacity for most wood chemistry or monomer based sugars. With some exception, ATR-FTIR models did perform reasonably strong for lignin, extractives, and xylose (Figs. 3 and 4). This was compatible to FT-NIR which also showed the highest predictive capacity for these three dependent variables.

In this research, ATR-FTIR and FT-NIR models were calibrated on measurements made from wet chemistry and HPLC. The RPD values of Holo-HPLC based models performed better than when the data was obtained from wet chemistry. This suggests that HPLC provides better accuracy and precision than wet chemistry based methods for sugar and hemicellulose estimation.

### 3.2.1. Interpretation of FT-NIR and ATR-FTIR models

Combining multivariate modeling with two instruments can provide additional insight in that similar band assignments for both ATR-FTIR and FT-NIR may be found for a given perturbation. Such agreement between models and instruments provides additional confidence that the right independent parameters were chosen for the model. Multiple instruments can further aid in the interpretation of key mechanisms that contribute to the prediction of a chemical constituent. Table 3 provides a summary of key band locations in both the near and mid infrared region in the prediction of wood chemistry. For most models, there was no clear trend as to which modeling technique worked better for interpretation. PCR appeared to work as well as PLS when assigning bands and making interpretations. This suggests that PCR is more comparable to PLS if the goal is to interpret the effect of underlying mechanisms on the  $y$  variable of interest. However, if ranking of statistical variables through statistical hypothesis testing is important, PCR may be preferred over PLS (Via, 2010; Via et al., 2013). During the prediction of density with ATR-FTIR, it was found that PCR worked better than multiple linear regression (Via, Fasina, & Pan, 2011). The current study suggests that PLS could be used as a substitute for PCR when interpreting models. But care should be taken since the loading bands can differ due to error during estimation (Via et al., 2014). This may be useful since many studies prefer PLS modeling due to better prediction diagnostics as witnessed in this study. In summary, it has been shown that when many studies in other scientific fields were observed, PCR generally outperformed PLS for interpretation while PLS was better for prediction of the  $y$  variable (Senter, 2008; Via et al., 2014).



**Table 3**

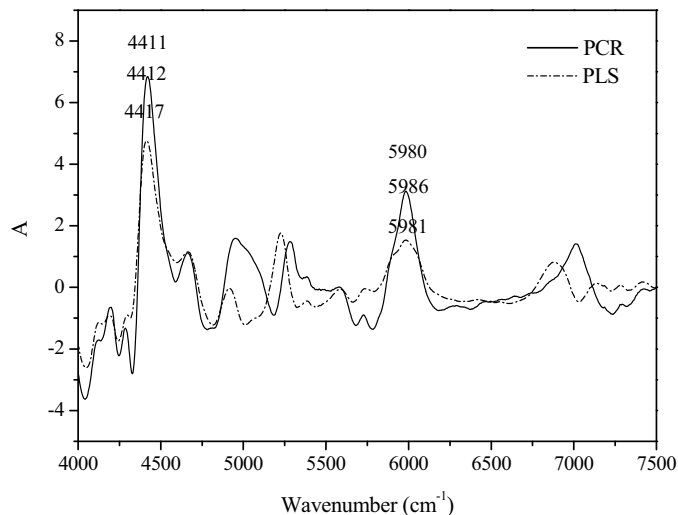
Significant wavenumber extracted from PLS and PCR modeling for the prediction of wood chemistry and monosaccharides for FT-NIR and ATR-FTIR.

| Chemical constituent | FT-NIR |      | FT-NIR band location | Band assignment                               | ATR-FTIR |      | ATR-FTIR band location | Band assignment                         |
|----------------------|--------|------|----------------------|-----------------------------------------------|----------|------|------------------------|-----------------------------------------|
|                      | PCR    | PLS  |                      |                                               | PCR      | PLS  |                        |                                         |
| Extractives          | 5205   | 5197 | 5220–5150            | O–H asym. str. + O–H def. of H <sub>2</sub> O | 1269     | 1269 | 1271                   | C–O                                     |
|                      | 5994   | 5961 | 5995                 | C–H str.                                      | 1506     | 1508 | 1510                   | C–C or C=C                              |
|                      | –      | –    | –                    | –                                             | 1636     | 1640 | 1633                   | –                                       |
| Lignin               | –      | –    | –                    | –                                             | 1733     | 1732 | 1730                   | C=O str                                 |
|                      | 4412   | 4417 | 4411                 | O–H str. + C–O str.                           | 833      | 833  | 835                    | O–H or C–H def.                         |
|                      | –      | –    | –                    | –                                             | 1374     | 1375 | 1375                   | –                                       |
| Cellulose            | 5986   | 5981 | 5980                 | C–H str.                                      | 1023     | 1011 | 1033                   | Aromatic C–H def. + C–O def. + C=O str. |
|                      | –      | –    | –                    | –                                             | 1124     | 1126 | 1116                   | –                                       |
|                      | –      | –    | –                    | –                                             | 1214     | 1211 | 1220                   | C–C + C–O + C=O str.                    |
| Hemicellulose        | –      | –    | –                    | –                                             | 1326     | 1323 | 1328                   | C=O str.                                |
|                      | –      | –    | –                    | –                                             | 1734     | 1731 | 1735                   | –                                       |
|                      | –      | –    | –                    | –                                             | 1505     | 1503 | 1510                   | C–C + C=C                               |
| Glucose              | 4412   | 4412 | 4405                 | O–H str. + C–O str.                           | 1234     | 1187 | 1235–1225              | O–H def.                                |
|                      | –      | –    | –                    | –                                             | 1019     | 1029 | 1120–1000              | Carb.                                   |
|                      | –      | –    | –                    | –                                             | 1059     | 1072 | 1051                   | C–O–C str.                              |
|                      | –      | –    | –                    | –                                             | 1263     | 1260 | 1267                   | C–H <sub>2</sub>                        |
|                      | –      | –    | –                    | –                                             | 1419     | 1423 | 1426                   | C–H + O–H bend.                         |
|                      | –      | –    | –                    | –                                             | 1507     | 1509 | 1509                   | C–C + C=C                               |
|                      | –      | –    | –                    | –                                             | 1633     | 1604 | 1639                   | Water                                   |
|                      | –      | –    | –                    | –                                             | 1026     | 1044 | 1025                   | C–O or C–C vibration                    |
|                      | –      | –    | –                    | –                                             | 1137     | 1137 | 1149                   | –                                       |
|                      | –      | –    | –                    | –                                             | 1221     | 1188 | 1203                   | C–H or O–H                              |
|                      | –      | –    | –                    | –                                             | 2922     | 2922 | 2913                   | –                                       |
|                      | –      | –    | –                    | –                                             | 1462     | 1456 | 1460                   | C–H <sub>2</sub> + O–C–H + C–C–H        |

sym.: symmetric, asym.: antisymmetric, str.: stretching vibration, bend.: bending vibration, def.: deformation vibration, carb.: carbohydrates.

In this study, for the prediction of extractives content, the C–H, C–C and C=C stretch were important loadings (Table 3). For lignin, the O–H and C–H bond were common to both instruments, while C=O was important for ATR-FTIR. For cellulose prediction, the O–H deformation was key functional groups identified by both techniques, but C–O was vital to FT-NIR. Hemicellulose quantification was perhaps the most different with the band assignment. The difference in models in the prediction of hemicellulose may be beneficial and suggests that both instruments might be combined to yield better predictive capacity. This may be important since, in this study, the prediction of hemicellulose was more difficult than other wood polymers with lower predictive capability. It is thus no surprise that these same differences between instruments were also observed during the modeling of glucose (Table 3).

It should finally be noted that the FT-NIR and ATR-FTIR bands listed in Table 3 were obtained through PCR and PLS modeling and may differ from some assignments found in the literature through similar or other methods. Fig. 5 shows an example a loading plot for PCR and PLS models for lignin content. Two bands were assigned to lignin from the literature (4411 and 5980 cm<sup>−1</sup>) (Schwanninger, Rodrigues, & Fackler, 2011). When the PCR and PLS models were built and loadings compared, for lignin, the models overestimated the wavenumber by as much as 6 cm<sup>−1</sup> (Fig. 5). When the loadings for all chemistry models were compared for FT-NIR, most values fell within  $\pm 5$  from common literature based assignments, but in several cases was as high as  $\pm 10$  (Table 3) and in a few cases as high as  $\pm 20$ . Similar errors for ATR-FTIR chemometric models were also present for the biopolymers explored in this study. This new finding suggests that consideration for model errors in loading estimates is important when assigning wavelengths to biopolymer concentration if one is to avoid misleading conclusions during band assignment or interpretation.



**Fig. 5.** A comparison of PCR and PLS loadings (raw spectra) for the prediction of lignin with FT-NIR spectroscopy. The wavenumber of two important bands were placed within the graph in which the top number represents values extracted from the literature, the middle number represents the loading value obtained through PCR and the bottom number represents that obtained through PLS.

#### 4. Conclusions

The chemical composition of extractive, lignin, cellulose, hemicellulose, holocellulose, glucose and xylose in hardwood were determined using wet chemistry and HPLC. Prediction models were built for these chemical compositional components using PLS and PCR with ATR-FTIR and FT-NIR spectra. It was found that using HPLC can get the more accurate holocellulose content measurements

than when using the wet chemistry methods. ATR-FTIR spectra can be used for screening purposes while FT-NIR often could be used for monitoring the quality of specific samples. PLS performed best during the prediction chemical composition with the best prediction models being extractive, lignin and xylose. PCR worked as good as PLS if interpretation of the model was important. It was also found that FT-NIR and ATR-FTIR models could be compared to gain additional insight into the effect of functional groups on chemical composition prediction.

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## Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.carbpol.2014.11.062>.

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